



CIRCUIT EDIT THROUGH COPPER POWER PLANE WITH HALOGEN

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TECHNOLOGY AND CIRCUIT EDIT

- **In an effort to enhance the performance of chip**
 - the IC industry made a switch in materials from aluminum to copper for use in metallization layers.
 - Since FIB are an integral part of *IC Debug, Fault-Isolation* and *low Yield Analysis*, an immediate problem was encountered when copper metal was milled in the FIB.
- **In general Cu process differs among manufactures, yields different results**
 - Therefore different FIB milling parameters are required.
- **Focused Ion Beam is tracking the evolution of IC interconnect material trends by developing**
 - application-specific chemistries.
 - circuit edit approaches
- **Continuous CE challenge**
 - Address both laterally and vertically scaling of dimensions
 - Coming soon new generation of Copper

CHALLENGES OF CIRCUIT EDIT

- **Which are the expectation on chemistries in CE?**
 - Enables different material deposition
 - Helps to improve removal while protecting surrounding materials.
 - Preferably chemistry must be passive until activated by the ion beam.
 - Spontaneous reaction is undesirable, exceptions can be contemplated:
 - XeF_2 etches spontaneously Si, but not SiO_2 (so it can be used to etch various dielectrics)
 - Heavy halogens such as Cl, Br, I etch very well Al, but spontaneously corrodes Cu
- **Copper metallization**
 - Slows down CE processes
 - Copper unlike Al has bigger grains
 - These grains etch at different rates, depending on grain orientation relative to ion beam
 - Solution to etch Copper
 - Not accelerate removal of Cu but decelerate removal of SiO_2 (protect the dielectric)
 - Oxidizing Cu
 - Cut in progressive tilt mode without/without gas

The image shows a scanning electron micrograph (SEM) of a porous, white, granular material. The material has a highly textured, interconnected structure. In the center of the image, the text "INVESTIGATION TEST 1" is overlaid in a bold, blue, sans-serif font. The text is reflected below it, creating a mirror effect. The background is a dark, uniform color.

INVESTIGATION TEST 1

E-Beam	Mag	Det	FWD	Tilt	Spot	Scan	5 μ m
2.00 kV	12.0 kX	TLD-S	5.116	52.0°	3	H 69.07	AGRATE-FA-TQeCSR

EDI CU-BULK RECIPE

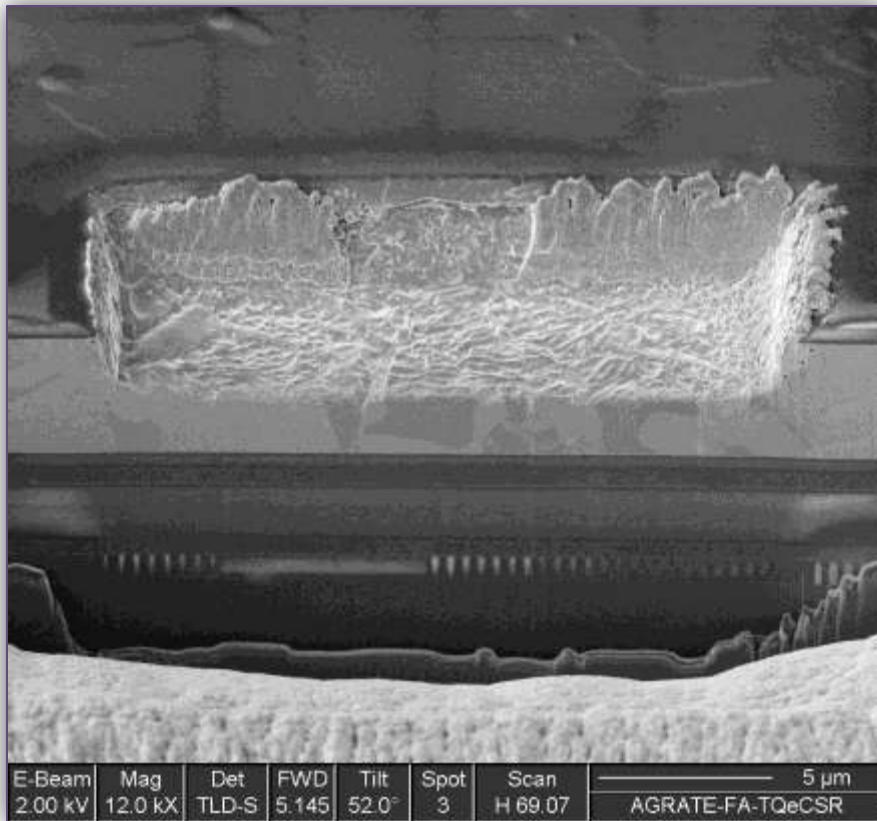
- **Goal**
 - Demonstrate the performance of the **Cu-Bulk EDI*** recipe
 - Compare results with the typical **Enhanced Copper Etch (Cu-II)** recipe
- **Recipe outline**
 - EDI (Ethylene diiodide) corrodes spontaneously exposed Cu and it is used to our benefit
 - Polycrystalline Cu gets amorphized and it loses Cu grain structure
 - Flat removal of Cu under controlled beam scanning parameters
 - Consideration on Ion Beam scanning parameters
 1. Provide enough time for chemistry to react spontaneously with Cu, to develop corrosion layer (chemistry pressure, box frame time, refresh time)
 2. Achieve a certain corroded Cu layer thickness that can be evenly removed by the ion beam (Beam current, dwell time, number of pixels)
 3. As long as corrosion layer is completely etched by each ion beam scan, the process works effectively.
 - Creating a corroded Cu layer and etching it instantly during FIB mill, provides very fast and planar copper removal.

TEST 1: EDI CU-BULK RECIPE QUALIFICATION

- **Setup of the investigation**
 - Open 2 different trenches with
 - *EDI CU-BULK* recipe
 - *Enhanced Copper Chemistry (Cu-II)* recipe
 - Milled box area is $18 \times 18 \mu\text{m}^2$
 - Cu power plane of $3,75 \mu\text{m}$ thickness
- **Cross section analysis at SEM**

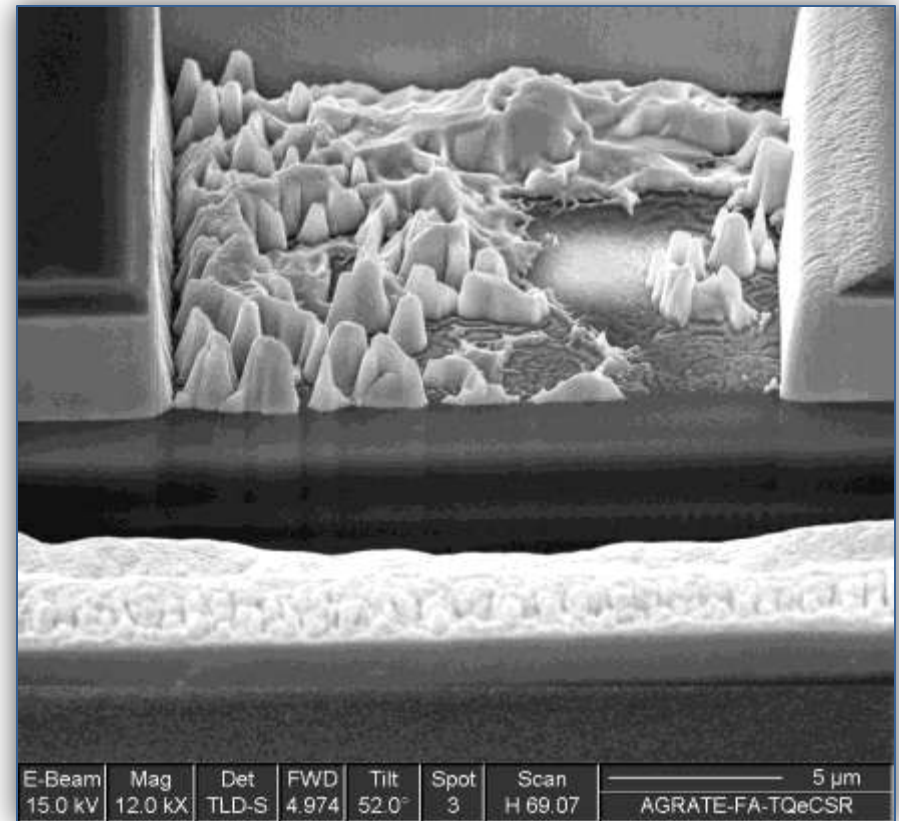
TEST 1: RECIPE COMPARISON AT HALF PROCESS

Bulk-EDI RECIPE



- 5 min recipe (half time)
- Flat removal progress

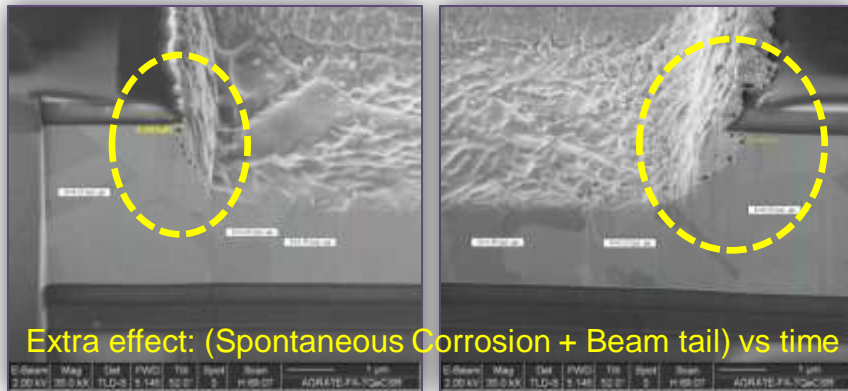
Cu-II RECIPE



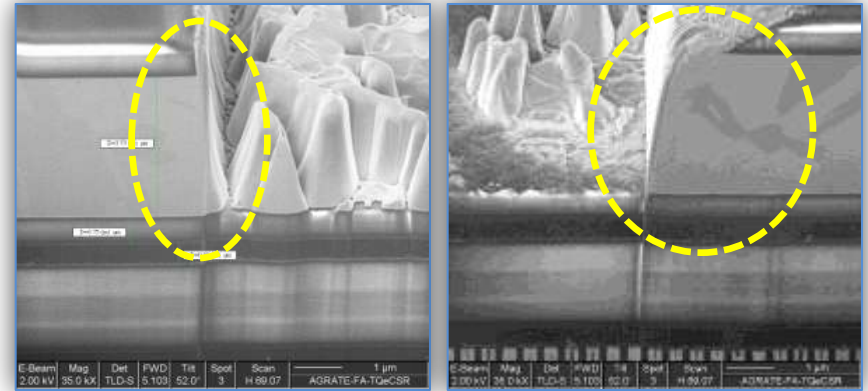
- 25min recipe (half time)
- Strong Topography, but high ILD protection

TEST 1: RECIPE COMPARISON AT HALF PROCESS

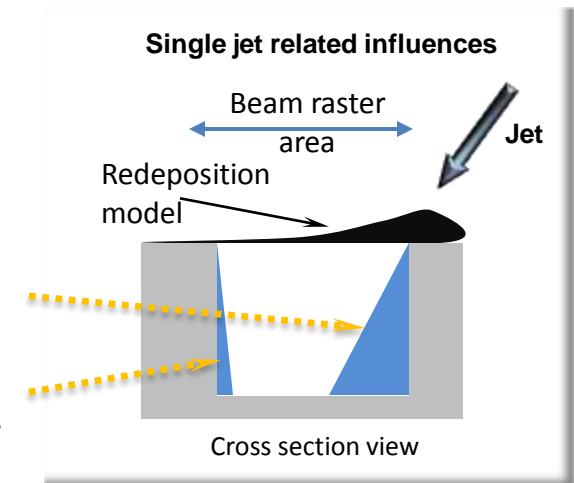
Bulk-EDI RECIPE



Cu-II RECIPE

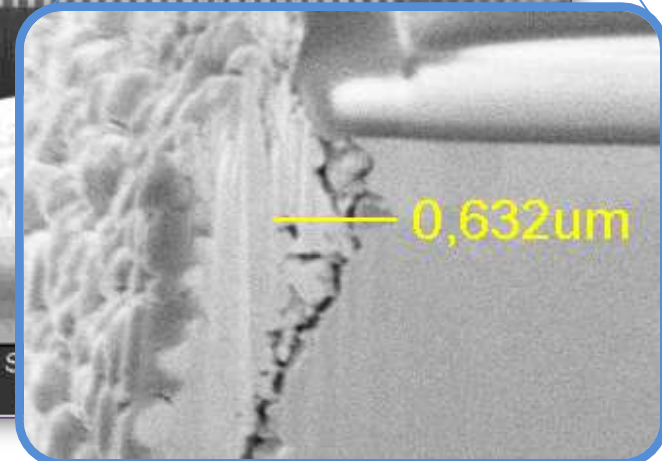
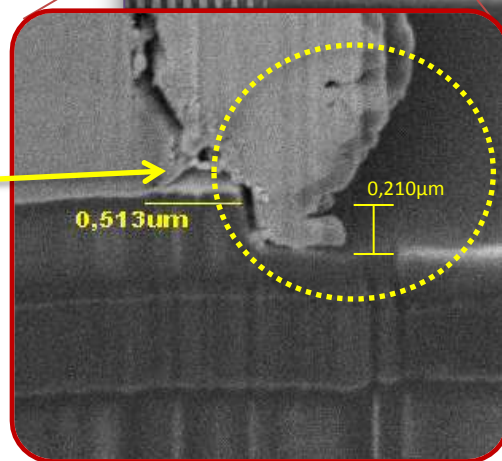
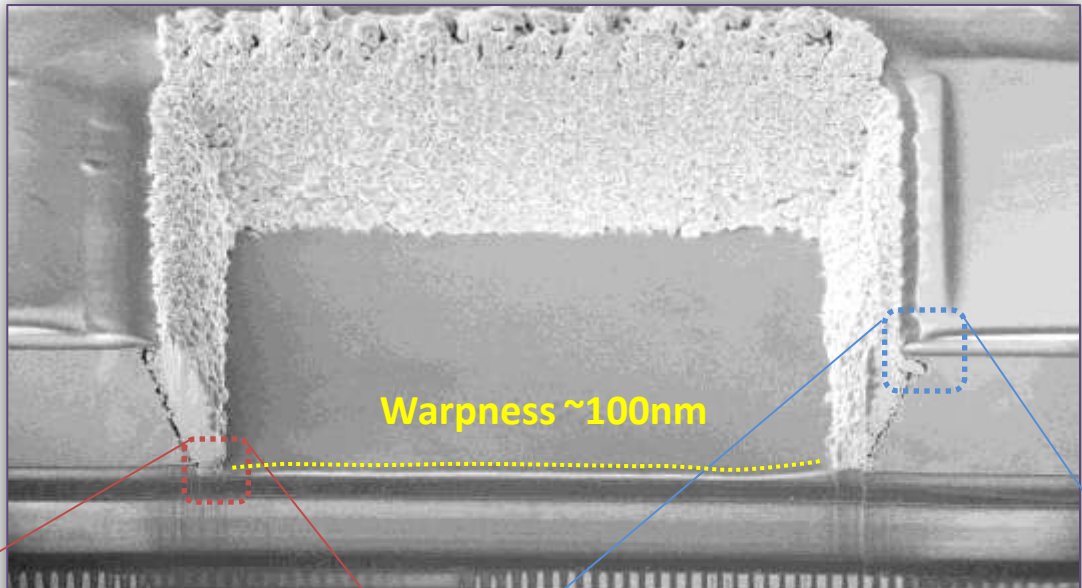


- Re-sputtered material is gas flux dependant and its redeposition appears on the jet side. The effect is influenced by gas flow rather than by beam raster.
- Side wall shadows gas flow. Reduced gas density does not give vertical wall and clean cut
- Higher gas provides steeper side walls and clean cut.



TEST 1: VIEW OF THE CUT WITH BULK-EDI RECIPE

- 10 min of recipe to remove 3,75 μm of Cu power plane
- Flat and clean Cu power plane removal
- Under the top passivation the sidewall overetch is $\sim 632\text{nm}$
- ILD overetch over the full trench is $\sim 210\text{nm}$ (ILD is 500nm)
- View of the amorphization progress above etched area



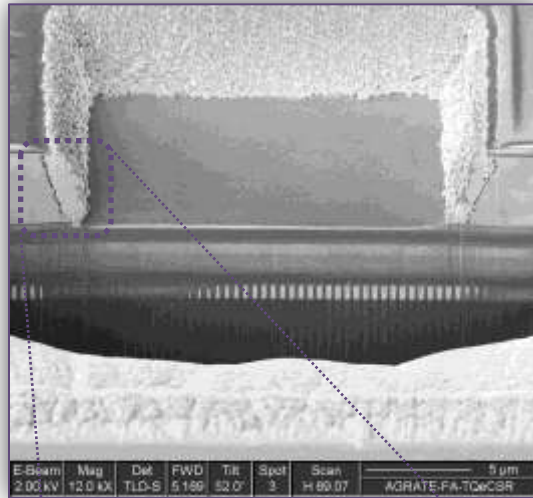
Tilt
52.0°

TEST 1: RECIPES COMPARISON AT THE END

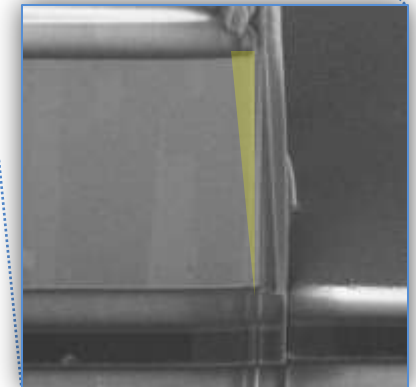
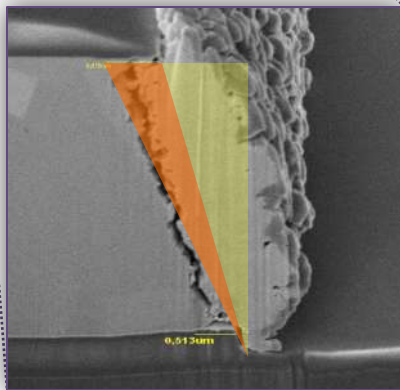
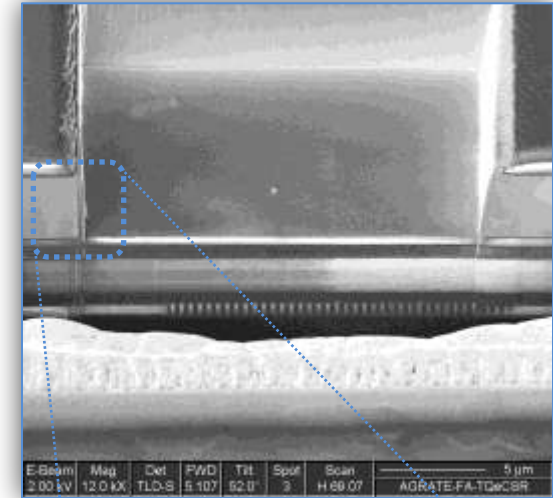
Bulk-EDI RECIPE

vs

Cu-II RECIPE



TIME	
10 min	45 min
BOX AREA	
Same bottom trench area	
ILD EXPOSED QUALITY	
Same flatness & warpsness	
Same ILD over etch	
BOX OVERTETCH	
~1,50μm	~500 nm



The image shows a scanning electron microscope (SEM) view of a porous, white, rectangular material. The material has a highly textured, granular surface. In the center of the material, the text "INVESTIGATION TEST 2" is overlaid in a bold, blue, sans-serif font. The text is reflected below it, suggesting a glossy or semi-transparent surface. The background is a dark, uniform grey.

INVESTIGATION TEST 2

E-Beam	Mag	Det	FWD	Tilt	Spot	Scan	5 μ m
2.00 kV	12.0 kX	TLD-S	5.116	52.0°	3	H 69.07	AGRATE-FA-TQeCSR

TEST 2: CONTAMINATION & CORROSION EVALUATION

What will happen during the CE to other exposed Cu areas?

EDI corrodes exposed Cu, even in areas out of FIB image field of view.

- **Goals**

- Define over the time the Cu corrosion by exposing it to an halogen compound (here EDI)
- Analyze parameters such as: Top surface, Corrosion & contamination progress, Artifacts
- Measure the impact of the corrosion during the CE process and also over the time.
 - The device with contaminated area has been kept under vacuum (2.5×10^{-7} Torr) for 10 days, in order to reduce external environment contamination.

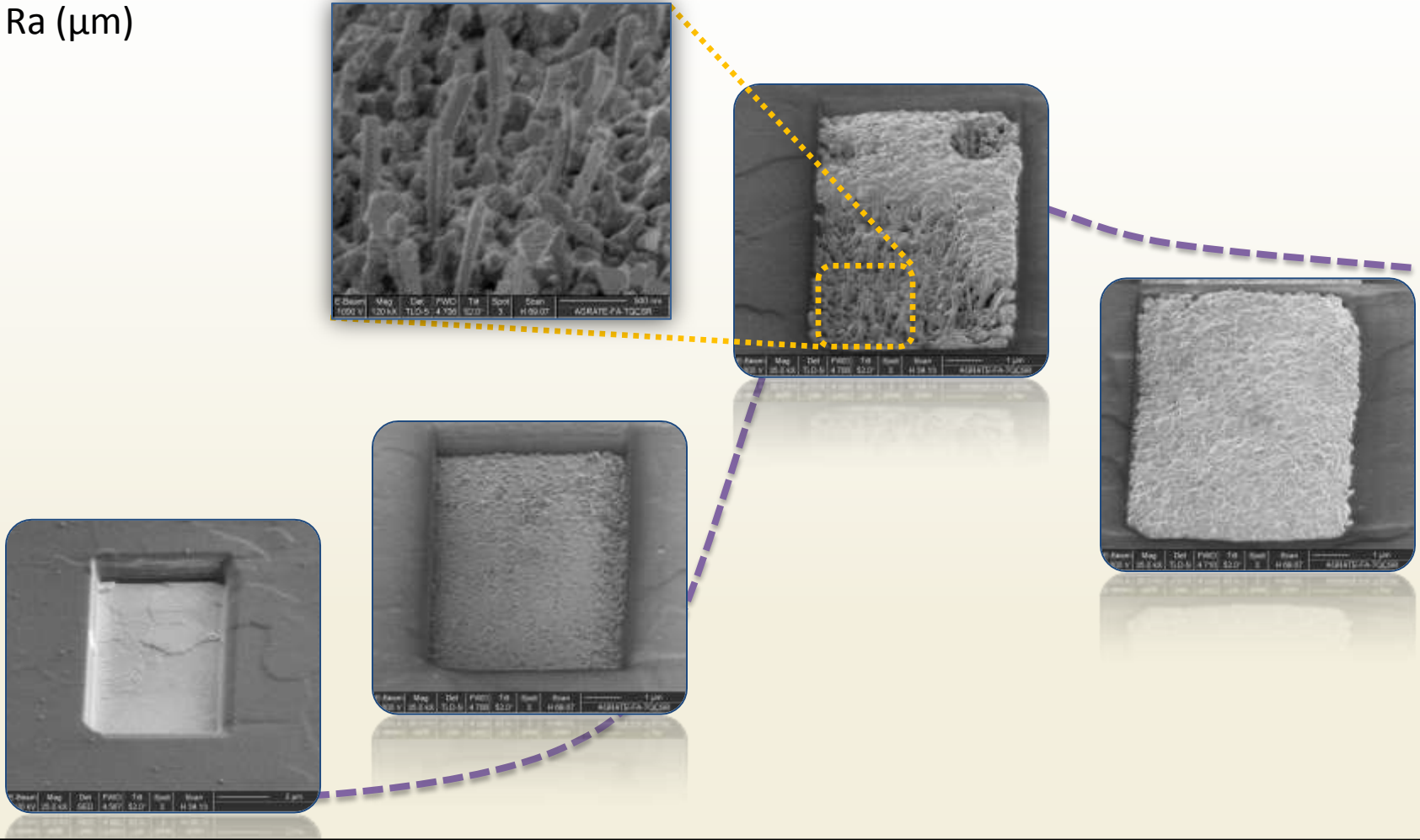
- **Setup**

- Expose 4 areas to the gas flow, respectively 0,1,5,10 minutes
- During the gas flow no beam rastering is applied
- Perform cross-section and SEM analysis
- Evaluate the contamination and corrosion amplitude in both lateral and vertical axis

TEST 2: -DAY 0- CU ROUGHNESS VS EDI EXPOSURE

Surface roughness

Ra (μm)



0

1

5

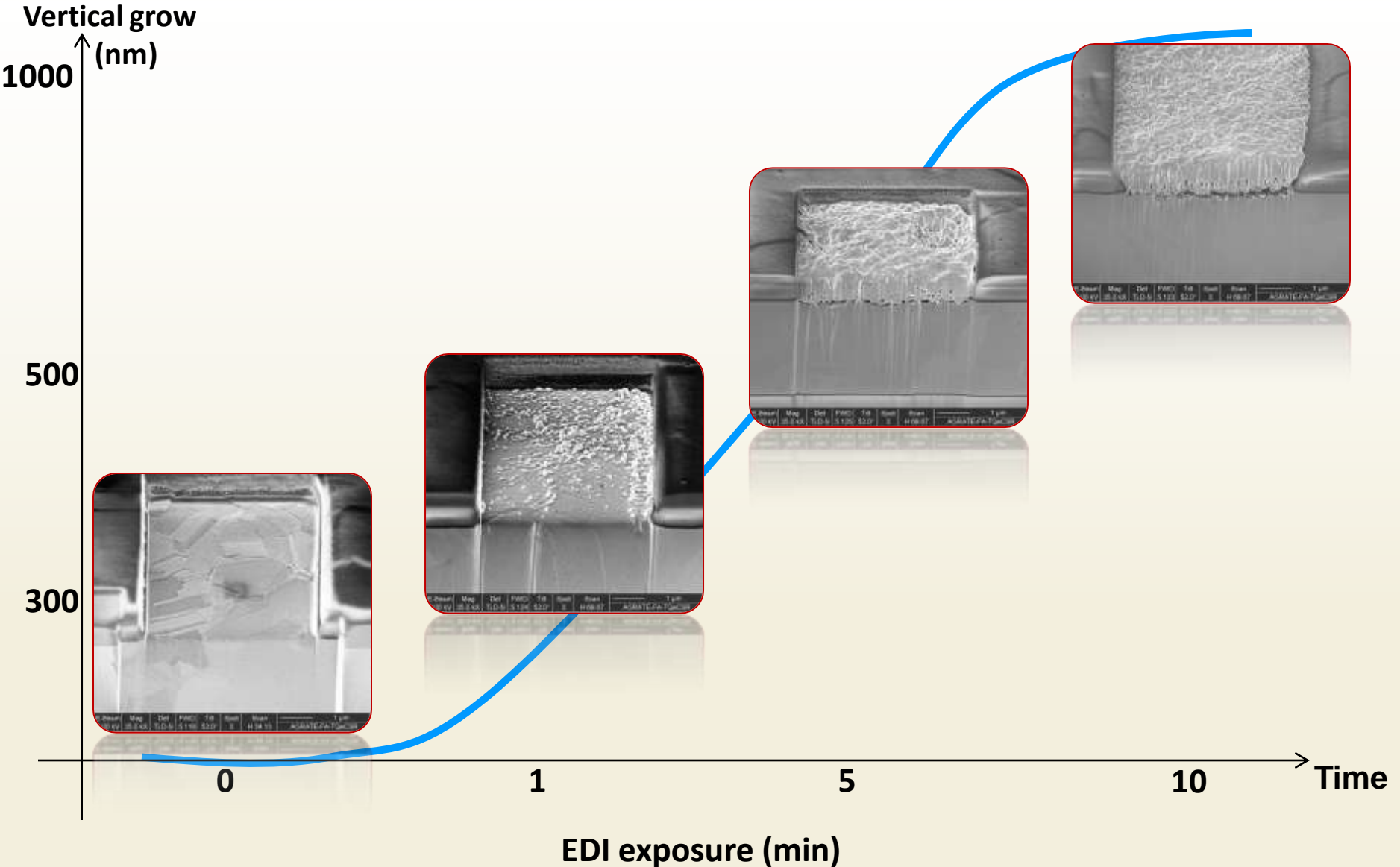
10

EDI exposure (min)

Time

Qualitative graph

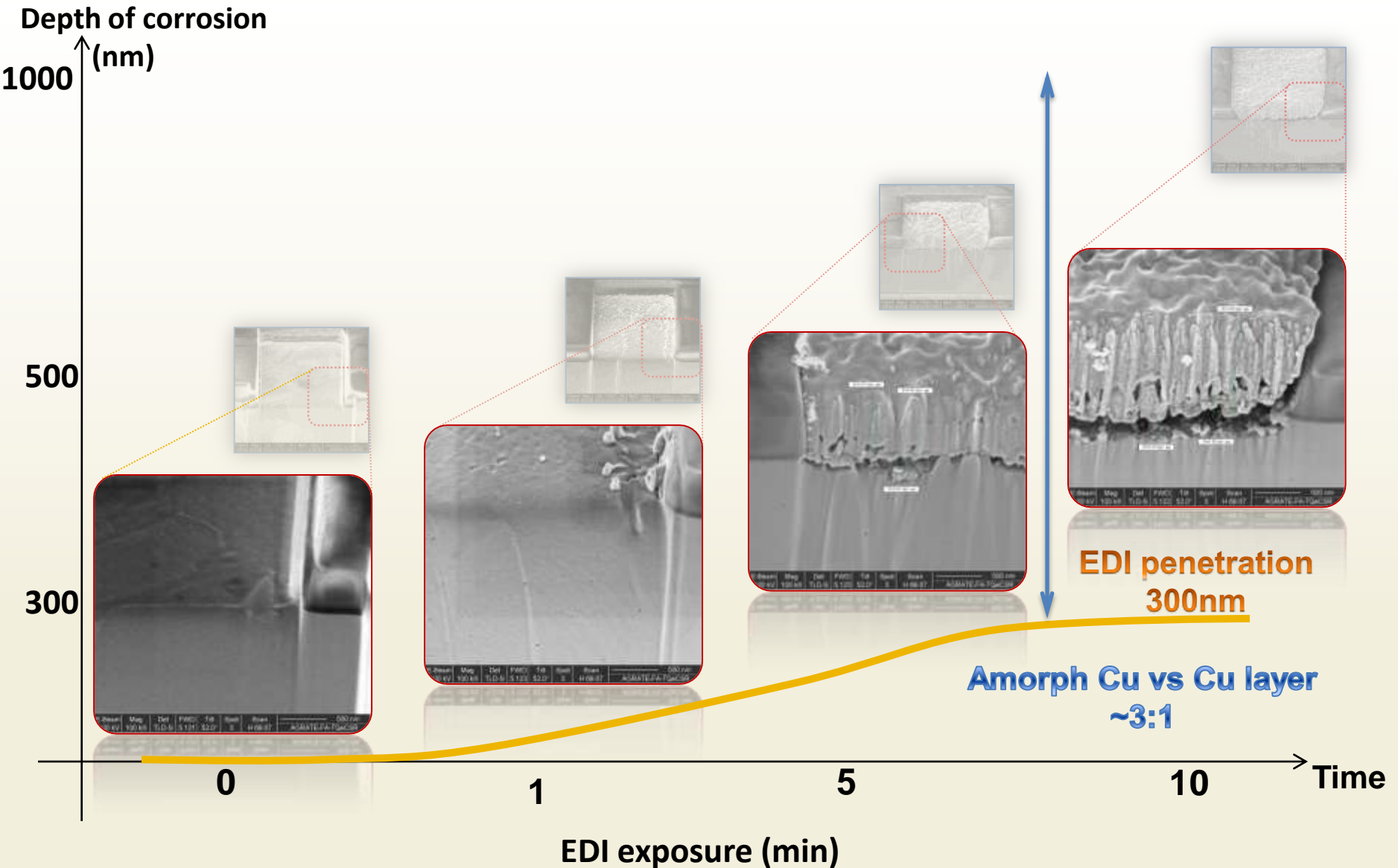
TEST 2: -DAY 0- CU AMORPHIZATION VS EDI EXPOSURE



Qualitative graph

EDI exposure (min)

TEST 2: -DAY 0- CU CORROSION DEEPNESS VS EDI EXPOSURE



Qualitative graph

TEST 2: -DAY 10- CU CORROSION DEEPNESS VS EDI EXPOSURE

Depth of corrosion

(nm)

1000

500

300

0

1

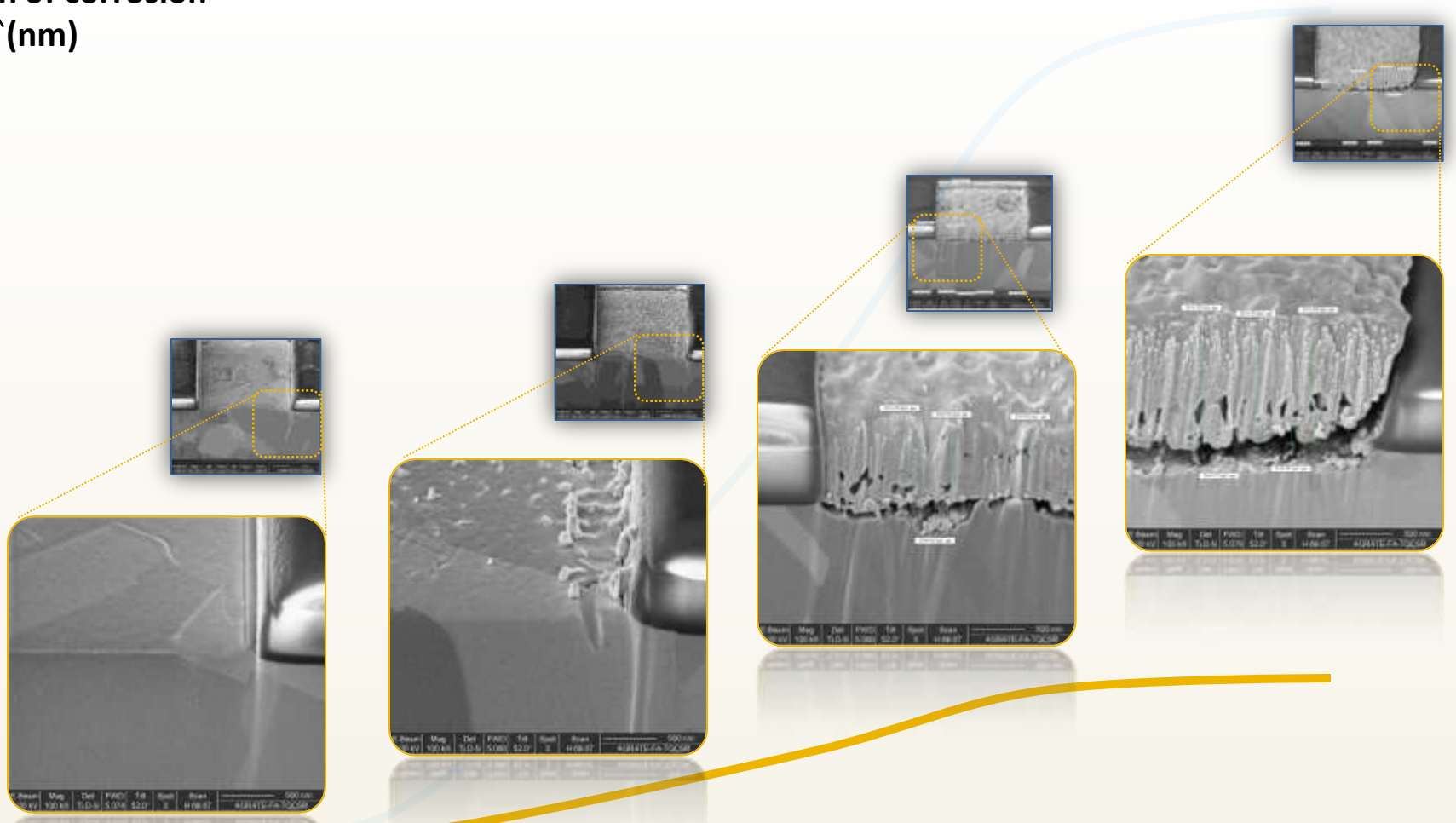
5

10

Time

EDI exposure (min)

Qualitative graph



CONCLUSION

- **Bulk-EDI RECIPE**

- The empirical results presented above qualify the very fast & planar Cu power plane etch rate
- The recipe is useful for large and thick Cu plane and is 4 time faster than Cu-II recipe.
- Turning strong crystallographic material in an amorphous one helps to enhance sputter yield.
- Works better if in presence of underlying thick dielectric
- The benefit of reducing the time to remove Cu power plane is surely an advantage for CE throughput
- Before introducing EDI into the FIB chamber for a Cu device edit, user must make sure all other exposed Cu areas are covered or it will be corroded as shown above. On thick metallization it is not an issue.

Acknowledge

- CAPITANIO Emanuele *STMicroelectronics*
- PEREGO Stefano *STMicroelectronics*

References

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